

Evaluation of Stability of the Medicinally Important Compounds in Coleus aromaticus plant by DFT method

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Abstract - Medicinal important compounds like were selected for their stability determination via binding energy calculation. The binding energy of these was determined by the Gaussian software 5.0. DFT were accomplished B3LYP and HF methods using three basis set be found of STO-3G, 3-21G & 6-31G. The DFT calculations of binding energy proved that Rosmarinic acid was recognized to be more stable among other seven compounds. Binding energy byST03-G (-1280.016), 32-1G (-1289.780) and 63-1G (-1296.548) were observed by B3LYP method. HF method binding energies were found to be STO-3G (-1272.803), 321G (-1282.173) and 63-1G (-1288.805). These results showed that among the medicinal compounds present in the Coleus aromaticus plant, Rosmarinic acid was found to have very good binding energy. Because of its good binding energy and stability it may find as potential medicine for the treatment of disease.

Keywords: Coleus aromaticus plant, DFT, B3LYP and HF method.

1. INTRODUCTION:

Coleus aromaticus is basically the family of Lamiaceae. The leaves of the plant are bitter, acrid and were being widely used traditionally for various purposes. The plant has been reported various biological properties [1]. Coleus aromaticus plants of medicinal uses are antiepileptic and antioxidant activities. The Coleus aromaticus leaves are also used for medicinal uses are antiepileptic and antioxidant activities. The Coleus aromaticus leaves are also used for treatment of cough, throat infection and nasal congestion [2-6]. Coleus aromaticus reported antimicrobial activity and oil against pathogenic and non-pathogenic fungi and bacteria [7]. It is also reported to possess remarkable diuretic property and food additive properties. Antioxidant activity is reported to be mainly due to rosmarinic acid, chlorogenic acid and caffeic acid. The essential oil of Coleus aromaticus has great anti-microbial activity on Gram negative as well as Gram positive bacteria. drug resistant microorganism, phytopathogenic microorganism and fungi [8]. It is also reported to possess therapeutic efficacy to treat malarial fever, chronic asthma, cancer, bronchitis,

helminthiasis, convulsions and epilepsy [9-10]. It has been reported that the essential oils extracted from these plants have potent activity against microorganisms [11]. In our present work, binding energies of eight constituents were computated using the B3LYP and HF methods by DFT approach.

2. Experimental Methods

2.1 Materials

The medicinal important compounds present in the Coleus aromaticus plant were selected for our work from the literature as given the structures below in Fig.1-8.





Fig -7: Chlorogenic acid Fig -8: Rosmarinic acid

DFT Methods

The GaussView 5.0 software was used to draw the structures of the compounds. Binding energy of the above compounds were determined using Gaussian software, binding energy by B3LYP and HF methods using three basis sets STO-3G, 3-21G and 6-31G [12].

Table-1 Binding energy of compounds in B3LYP method

S. No	Compounds name	Basis sets		
		STO-3G	3-21G	6-31G
	Chavicol	-418.681	-421.690	-423.912
2.	Carvacrol	-458.938	-462.168	-464.592
3.	Eugenol	-531.683	-535.572	-538.381
4.	Coumaric acid	-565.735	-569.999	-573.004
5.	Caffeic acid	-639.883	-644.788	-648.184
6.	Salvigenin	-1131.982	-1140.441	-1146.406
7.	Chlorogenic acid	-1279.691	-1289.660	-1296.403
8.	Rosmarinic acid	-1280.016	-1289.780	-1296.548

Table- 2 Binding energy of compounds in HF method

S. No	Compounds name	Basis sets		
		STO-3G	3-21G	6-31G
	Chavicol	-416.032	-418.942	-421.126
2.	Carvacrol	-456.037	-459.125	-461.506
3.	Eugenol	-528.441	-532.180	-534.936
4.	Coumaric acid	-562.433	-566.546	-569.492
5.	Caffeic acid	-636.264	-640.978	-644.304
6.	Salvigenin	-1125.510	-1133.613	-1139.461
7.	Chlorogenic acid	-1272.635	-1282.135	-1288.736
8.	Rosmarinic acid	-1272.803	-1282.173	-1288.805

Result and Discussion 2.2 DFT Calculation 2.2.1 B3LYP Method

The binding energy determination by three basis sets by B3LYP method as given in Table-1 showed that whereas other compounds like Chavicol and Carvacrol were found have poor binding energy (-418.681 and - 458.938 a.u. by STO-3G), (-421.690 and -462.168 a.u. by 3-21G) and 6-31G method (-423.912 and -464.592 a.u.). The compounds Eugenol, Coumaric acid, Caffeic acid and Salvigenin were found to be less stable as per the binding energy in the range STO-3G method (-531.683, -565.735, -639.883 and -1131.982) a.u. 3-21G method (-535.572, -569.999, -644.788 and -1140.441) a.u. & 6-31G method (-538.381, -573.004, -648.184 and -1146.406) a.u. Chlorogenic acid and Rosmarinic acid were found to have good binding energy STO-3G (-1279.691 & -1280.016 a.u.), 3-21G (-1289.660 & -1289.780 a.u.) 6-31G (-1296.403 & -1296.548 a.u.) respectively.

HF Method

The binding energy calculated to the above compounds given in **Table-2** for the three basis sets. Binding energy calculated by STO-3G basis sets were -416.032, -456.037, -528.441, -562.433, -636.264, -1125.510, -1272.635 & -1272.803 a.u. for Chavicol, Carvocrol, Eugenol, Coumaricacid, Caffeicacid, Salvigenin, Cholrogenicacid and Rosmarinic acid compounds. 3-21G basis sets predicted -418.942, -459.125, -532.180, -566.546, - 640.978, -1133.613, -1282.135 & -1282.173 a.u. and 6-31G basis sets showed -421.126, -461.506, -534.936, -569.492, -644.304, -1139.461, -1288.736 & -1288.805 a.u. as binding energies for the above compounds as given in the same order.

4. Conclusion

Among the eight compounds selected for the determination of binding energy to predict their stability. The Rosmarinic acid was found to have good binding energy by B3LYP and HF methods Chavicol and Carvacrol were found to be very poor binding energy values less than -500 a.u. The compounds Eugenol, Coumaric acid and Caffeic acid were found to be less stable as per the binding energy values less than -1000 a.u. The Rosmarinic acid was found to have very good binding energy by both the B3LYP method (-1280.016, -1289.780 & -1296.548 a.u.) and by HF method (-1272.803,-1282.173 & -1288.805a.u.) compared to other compounds. Prediction of binding energy revealed that Rosmarinic acid was found to be more stable than Chavicol, Carvocrol, Eugenol, Coumaricacid, Caffeic acid, Salvigeinin, and Cholrogenic acid. Hence Rosmarinic acid can be used as stable drug compared to other compounds.

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