Analysis of Cooling Performance of Automobile Radiator Using

Nano-fluid TiO2 and water

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_____***__ **ABSTRACT-** The main objective of this paper is to analyze the heat carrying capacity of the coolant used in the radiator of an automobile by using Nano-particle titanium oxide (Tio₂). The heat carrying capacity of the coolant can be increased as compared to that of conventional coolant (water) by adding Tio₂ as a coolant [1]. The experiment was conducted by varying the concentrations of the Nano-particle TiO_2 in water at three different flow rates. The concentration of the TiO_2 varies from 0.1% to 0.2% and the flow rate of the Nanocoolant has been kept in between 2lt/min to 4lt/min. The observations were made within a temperature range of 50°c -30°c to analyze the effect of fluid inlet temperature on the heat transfer coefficient. The test rig consist of radiator of capadity1000cc, ducts made up of Galvanized iron sheet of thickness 20 gauge, a pump employed for the continuous circulation of the coolant, 4 temperature sensors[2] to check temperature at different section of the radiator(inlet, outlet, wall1 i.e. front side, wall2 i.e. rear side), a hand blender used for mixing of Nano-particles with water and an immersion rod to increase the temperature of the coolant to $50^{\circ}c$. The best results were obtained when the coolant was circulated at flow rate of 3.4lt/min at 0.15%volume concentration of TiO2 in water. Keywords- Heat transfer coefficient, Nano-fluid, Convection,

Keywords- Heat transfer coefficient, Nano-fluid, Convection conduction, nusselt number, and Reynolds number.

1. INTRODUCTION

Under heavy loading, the engine of the automobile needs more power and hence cause the temperature of the engine to rise. For providing proper cooling to the engine, the heat dissipation is necessary and it is increased by adding nanoparticles in the coolant. Radiator has fins for heat dissipation through conduction and a fan blows air towards the radiator for heat dissipation through forced convection. [3]

Water was used as conventional fluid in radiator. The heat carrying of the water is not appropriate for proper cooling of the engine. For improved efficiency of the coolant, Nanoparticle (TiO_2) are added in the water at different concentrations. The nano fluid so prepared is circulated in the test rig at different flow rates and concentration of the nano particle in water. The readings were taken to find the flow rate and concentration at which max heat transfer coefficient was obtained. [4]

2. SOLUTION PROPOSED

By adding nano particles in the water at different volume concentrations and flow rates, an improved results in Reynolds no, nusselt no, and heat transfer coefficient is obtained. Two methods are deployed for obtaining an enhanced result in the heat carrying capacity of the coolant. The first method is that the flow rate is varied to obtain maximum heat transfer rate of the coolant. The second method employs change in concentration of the nano particle in water to check for the ideal concentration for maximum efficiency of heat transfer rate. If both the methods are used simultaneously then the results for the heat transfer rate are more efficient than that obtained with water alone. [5]

3. EXPERIMENTAL SETUP

The test rig constructional features are explained with the help of a CAD model as shown in figure 1. It focuses on observing the changes in heat transfer rate after addition of the nano-particle and obtaining the most efficient heat transfer rate of the nano-fluid. The test rig consist of a radiator, GI sheet duct[6], PT-100 temperature sensors, immersion rod for heating, a bucket as a reservoir for the nano-coolant and a hand blender for mixing of water and nano-particle.





Flow Chart of proposed solution:

The flow chart of the solution is demonstrated in the figure 2.



FIGURE 2: Experimental Setup of the Proposed Solution.

- 1. The Nano-coolant is prepared by mixing TiO_2 particles in three different volume concentrations (0.1%, 0.15%, and 0.2%) in water (10 Litres).
- 2. The Nano-fluid thus prepared is stored in container (bucket). An immersion rod is used to increase the temperature of the coolant to 50° c and a hand blender is employed to mix the particles of Tio₂ with water.
- 3. The flow rate of the coolant is varied by using a control valve from 2.7l/min to 4l/min.
- 4. The Nano-fluid is then pumped to the radiator inlet by using hoses (flow lines).
- 5. The air is then forced by using a fan which is operated by 220 volt D.C. battery.
- 6. The coolant is allowed to flow through the 1000cc radiator which is enclosed in duct of G.I sheet of thickness20 gauge.
- 7. The coolant when flows through the tubes of the radiator cools down (temperature drop) due to forced convection and conduction process.
- 8. The temperature drop is recorded and graphs are plotted to analyze maximum heat transfer rate.

4. EXPERIMENTAL CALCULATIONS

Nusselt no, Reynolds no and heat transfer coefficient are the three parameters that are used for observing the heat transfer rate of the coolant. The following terms are calculated with the help of specific heat capacity[7] and density[8]. The formulas of the same are given.

The specific heat capacity of the nano- coolant is calculated as:

 $C_{nf} = \frac{(1-\phi)(\rho C)_{W} + \phi(\rho C)p}{(1-\phi)(\rho C)_{W} + \phi\rho_{p}}$ (1)

C_{nf}= specific heat capacity of the nano coolant(J/Kgk)

 $C_{w=}$ specific heat capacity of water(4180 J/kgk)

C_{p=} specific heat capacity of TiO₂(683 J/kgk)

φ= Nano Fluid volume concentration % (at three different values 0.1%,0.15% and 0.2%)

$$\rho_{nf} = \text{density of nano fluid(kg/m^3)}$$

Density of the nano-fluid (ρ_{nf}) is calculated by using the given formula:-

$$\rho_{\rm nf} = \varphi \rho_{\rm p} + (1 - \varphi) \rho_{\rm w}$$
⁽²⁾

Where $\rho_w = \text{density of water}(1000 \text{kg/m}^3)$

$$\rho_{\rm p=}$$
 density of TiO₂(3970 kg/m³)

Heat transfer rate is calculated by given equation:

$$Q = mC_{nf}(T_{in} - T_{out})[9]$$
(3)

Where, m = mass flow rate of the nanocoolant(Kg/min)

T_{in}= Inlet temperature (°c)

T_{out}= outlet temperature(°c)

Heat transfer coefficient is calculated by using the given equation

From Newton's law of cooling:

$$Q = hA(T_b - T_s)$$
(4)

Where Q is the heat transfer rate(watt)

A is the surface area of the tube of radiator (217cm²)

 T_b is the bulk temperature (°c) which is calculated by taking the average of T_{in} and T_{out}

$$T_{b} = \frac{T_{in} + T_{out}}{2}$$
(5)

 T_s is the average wall temperature of the radiator measured from various transverse and longitudinal locations of radiator(°c)

$$h_{exp} = \frac{mC_{nf}(T_{in} - T_{out})}{nA(T_b - T_s)}$$
(6)

Where n = number of tubes (50)

The average Nusselt number can be calculated as:

$$N_u = \frac{hD_h}{k}$$

 $Where \,\, D_{h^{\pm}} \, hydraulic \, diameter [10] \, of the \, tube \, and \, is \\ calculated \,\, as$

(7)

$$D_{h} = \frac{4\left[\frac{\pi d^{2}}{4} + (D-d)d\right]}{\pi d + 2(D-d)}$$
(8)

D and d are the width and height of radiator tube.

Here d=1.8 mm; D=15.5mm.

Finally the Reynolds number can be calculated as:

 $R_{e=} \frac{\rho_{nf} v D_{h}}{\mu}$

(9)

Wherepnf =density of nano-fluid(kg/m³)

 μ = dynamic viscosity of the nano coolant(

v= Fluid velocity(m/s)

5. Observations:-

5.1: Nano Fluid (TiO₂+H₂O)

Table 1 shows the observations for TiO_2 based nano-fluid. Heat transfer coefficient (h), Reynolds number (Rn) and Nusseltnumber (Nu) are analyzed and shown in table 1. The same is plotted in figure 3. The readings were taken at three different flow rates(2.27lt/min, 3.4lt/min, 4lt/min) and concentrations (0.1%,0.15%,0.2%). The best result was obtained at a flow rate of 4lt/min and 0.15% volume concentration.

Table 1: Analysis of TiO₂based radiator coolant at different flow rates

| φ (%) | Flow rate(lt/min) | h (w/m²k) | Nu | Re |
|--------------|-------------------|-----------|------|------|
| | 2.27 | 40.84 | 1.3 | 5316 |
| 0.1 | 3.41 | 107 | 2.76 | 5316 |
| | 4 | 94.76 | 3.11 | 5316 |
| | 2.27 | 93.04 | 2.23 | 3272 |
| 0.15 | 3.41 | 140 | 3.36 | 3272 |
| | 4 | 117 | 2.8 | 3272 |
| | 2.27 | 90.61 | 1.8 | 2373 |
| 0.2 | 3.41 | 123 | 2.02 | 2373 |
| | 4 | 100 | 2.3 | 2373 |



Figure 3: Representing the variations in Reynolds number vs Nusselt number at different flow and concentrations for TiO_2

At four different junctions i.e. inlet, outlet, wall1, wall2 the temperature variation is observed with change in time at different flow rates and concentrations.

The table 2 and figure 4 represents the reading of temperature and time for TiO₂at 2.7 l/min flow rate and 0.10% volume concentration at regular time intervals.

Table 2: Temperature and Time forTiO₂at 2.7 l/m and 0.10 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 35.2 | 29.3 | 40.7 |
| 2.5 | 46.9 | 32.7 | 28.2 | 37.8 |
| 5 | 42.2 | 30.8 | 27.1 | 35.2 |
| 7.5 | 38.9 | 29.1 | 26 | 32.8 |
| 10 | 36.7 | 27.8 | 25 | 30.9 |
| 12.5 | 34.7 | 26.6 | 24.3 | 29.4 |
| 15 | 32.7 | 25.3 | 23.5 | 27.5 |
| 17.5 | 31 | 24.3 | 22.8 | 26.6 |
| 19.5 | 30 | 23.9 | 22.4 | 25.9 |



Figure 4: Temperature vs Time for TiO₂at 2.7 l/m and 0.10 % Volume Concentration

The table 3 and figure 5 represents the reading of temperature and time for TiO_2 at 3.4 l/min flow rate and 0.10% volume concentration at regular time intervals.

Table 3: Temperature and Time for TiO2 at 3.4 l/m and0.10 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 43.1 | 41.8 | 47.9 |
| 2.5 | 47.9 | 35.8 | 34.7 | 38.9 |
| 5 | 42.5 | 32.5 | 31.3 | 35.2 |
| 7.5 | 38.3 | 29.4 | 28.6 | 32.1 |
| 10 | 34.7 | 26.8 | 26.4 | 29.5 |
| 12.5 | 32.2 | 25.5 | 24.9 | 27.6 |
| 14.54 | 30 | 24.2 | 23.6 | 26.1 |

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Figure 5: Temperature vs Time for TiO_2 at 3.4 l/m and 0.10 % Volume Concentration

The table 4 and figure 6 represents the reading of temperature and time for TiO_2 at 4 l/min flow rate and 0.10% volume concentration at regular time intervals.

Table 4: Temperature and Time for Al2O3 at 4 l/m and0.10 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 37.4 | 36.8 | 43 |
| 2.5 | 45.5 | 34.7 | 34.3 | 39.3 |
| 5 | 41.8 | 32.1 | 32.1 | 36.1 |
| 7.5 | 38.3 | 30.1 | 30.1 | 33.5 |
| 10 | 35.5 | 28.1 | 28.2 | 31.2 |
| 12.5 | 33.2 | 26.6 | 26.7 | 29.4 |
| 15 | 31.2 | 25.2 | 25.5 | 27.8 |
| 16.47 | 30 | 24.6 | 24.9 | 26.8 |



Figure 6: Temperature vs Time for TiO_2 at 4 l/m and 0.10 % Volume Concentration

The table 5 and figure 7 represents the reading of temperature and time for TiO_2 at 2.7 l/min flow rate and 0.15% volume concentration at regular time intervals.

Table 5: Temperature and Time for $TiO_2\,at\,2.7$ l/m and 0.15 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 39.1 | 38.3 | 40.9 |
| 2.5 | 44.4 | 35.3 | 35 | 37.7 |
| 5 | 41.3 | 32.9 | 32.5 | 34.5 |
| 7.5 | 38.5 | 31.1 | 31 | 32.4 |
| 10 | 36.2 | 29.5 | 29.3 | 30.8 |
| 12.5 | 34.1 | 27.9 | 28.1 | 29.1 |
| 15 | 32.2 | 26.7 | 26.7 | 27.7 |
| 18.34 | 30.1 | 25.1 | 25.5 | 26.2 |



Figure 7: Temperature vs Time for TiO₂ at 2.7 l/m and 0.15 % Volume Concentration

The Table 6 and figure 8 represents the reading of temperature and time for TiO_2 at 3.4 l/min flow rate and 0.15% volume concentration at regular time intervals.

Table 6: Temperature and Time for TiO_2 at 3.4 l/m and 0.15% Volume Concentration

| Time, Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|------------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 35 | 35 | 40.3 |
| 2.5 | 42.7 | 31.3 | 33.2 | 34.9 |
| 5 | 38 | 28.9 | 30.9 | 31.6 |
| 7.5 | 34.3 | 26.7 | 28.4 | 28.9 |
| 10 | 30.9 | 24.2 | 25.8 | 25.8 |
| 11 | 30 | 23 | 25 | 25 |



Figure 8: Temperature vs Timefor TiO₂ at 3.4 l/m and 0.15 % Volume Concentration

The table 7 and figure 9 represents the reading of temperature and time for TiO_2 at 4 l/min flow rate and 0.15% volume concentration at regular time intervals.

Table 7: Temperature and Time for $TiO_2 \mbox{ at } 4 \mbox{ l/m}$ and 0.15 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 39.8 | 40.9 | 43.1 |
| 2.5 | 45.2 | 36 | 37.3 | 39.1 |
| 5 | 41.9 | 33.2 | 34.3 | 35.9 |
| 7.5 | 37.8 | 31.1 | 31.9 | 33.9 |
| 10 | 35.1 | 29.9 | 29.9 | 30.9 |
| 12.2 | 32.8 | 27.4 | 28.2 | 29.2 |
| 14.3 | 30 | 26 | 26.6 | 27.5 |

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18

14

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Outlettemp Walltemp1 Walltemp2 Outlettemp

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Figure 9: Temperature vs Time for TiO₂ at 4 l/m and 0.15 % Volume Concentration

The table 8 and figure 10 represents the reading of temperature and time for TiO_2 at 2.7 l/min flow rate and 0.20% volume concentration at regular time intervals.

Table 8: Temperature and Time for TiO2 at 2.7 l/m and0.20 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 48 | 33.9 | 37.2 | 38.2 |
| 2.5 | 43.3 | 31.8 | 34.1 | 34.8 |
| 5 | 40.9 | 30.8 | 32.5 | 33.3 |
| 7.5 | 36.8 | 28.9 | 30 | 30.5 |
| 10 | 34.1 | 27.3 | 28.3 | 28.6 |
| 12.5 | 32.2 | 26.2 | 27.1 | 27.3 |
| 15 | 30.5 | 25.2 | 26.2 | 26.2 |
| 16 | 28 | 24.9 | 25.8 | 25.8 |



Figure 10: Temperature vs Time for TiO_2 at 2.7 l/m and 0.20 % Volume Concentration

Table 9 and figure 11 represents the reading of temperature and time for TiO_2 at 3.4 l/min flow rate and 0.20% volume concentration at regular time intervals.

Table 9: Temperature and Time for $TiO_2\,AT\,3.4\,l/m$ and 0.20 % Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 37.4 | 36.4 | 41.8 |
| 2.5 | 45.3 | 35.3 | 35.3 | 38.3 |
| 5 | 40.9 | 32.4 | 33 | 35 |
| 7.5 | 37.9 | 30.5 | 31 | 32.6 |
| 10 | 35.1 | 28.8 | 28.5 | 30.5 |
| 12.5 | 32.8 | 27.3 | 27.9 | 28.5 |
| 15 | 30.9 | 25.8 | 26.4 | 27 |
| 16.1 | 30 | 25.4 | 25.9 | 26.5 |



Figure 11: Temperature vs Time for TiO₂ at 3.4 l/m and 0.20 % Volume Concentration

The table 10 and figure 12 represents the reading of temperature and time for TiO_2 at 4 l/min flow rate and 0.20% volume concentration at regular time intervals.

Table 10:Temperature and Time for TiO_2 at $4\,l/m$ and 0.20~% Volume Concentration

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 49 | 36 | 33.6 | 41.1 |
| 2.5 | 43.2 | 32 | 32 | 36.3 |
| 5 | 38.9 | 29.1 | 29.1 | 33.5 |
| 7.5 | 35.3 | 28.4 | 28.4 | 30.5 |
| 10 | 32.3 | 27.1 | 27.1 | 28.2 |
| 12.5 | 30 | 25.3 | 25.3 | 26.4 |
| 13.46 | 29 | 24.5 | 24.5 | 25.5 |



Figure 12: Temperature vs Timefor TiO₂at 4 l/m and 0.20 % Volume Concentration

By comparing the readings and observations from table two to table ten, it is found that the heat transfer rate is max at a flow rate of 3.4lt/min and a volume concentration of 0.15%.

5.2: Water (H₂O)

The heat transfer rate of distilled water is observed and analyzes at three different flow rates (2.27lt/min, 3.4lt/min, 4lt/min). The temperature variation is observed with change in time at different flow rates four different junctions i.e. Inlet, outlet, wall1, wall2.

The table 11 and figure 13 represents the reading of temperature and time water at 2.7lt/min flow rateat regular time intervals.

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Table 11: Temperature and Time for waterat 2.7lt/m

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 38.5 | 38.5 | 40 |
| 2.5 | 47 | 36.5 | 37.8 | 38.1 |
| 5 | 42.3 | 33.8 | 35.1 | 35.2 |
| 7.5 | 38.7 | 31 | 32.2 | 32.3 |
| 10 | 35.7 | 28.9 | 30 | 30 |
| 12.5 | 32.8 | 27 | 28.1 | 27.8 |
| 15 | 31.2 | 22.7 | 21.7 | 23.5 |
| 17.5 | 30 | 23.1 | 22.2 | 24.5 |



Figure 13: Temperature vs Time for waterat 2.7lt/min

The table 12 and figure 14 represent the reading of temperature and time water at 3.4 lt/min flow rateat regular time intervals.

Table 12: Temperature and Time for waterat 3.4lt/m

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 40.3 | 43 | 42.9 |
| 2.5 | 44.4 | 36.3 | 38.3 | 38.9 |
| 5 | 39.9 | 32.8 | 34.8 | 35 |
| 7.5 | 36.3 | 30.1 | 31.8 | 32 |
| 10 | 33.3 | 27.8 | 29.3 | 29.4 |
| 12.5 | 30.5 | 27.1 | 27.2 | 25.8 |
| 13.5 | 30 | 25.2 | 26.7 | 26.8 |



Figure 14: Temperature vs Time for waterat 3.4lt/min

The table 13 and figure 15represent the reading of temperature and time water at 4 lt/min flow rateat regular time intervals.

Table 13: Temperature and Time for waterat 4lt/m

| Time(Min) | Inlet temp(°c) | Wall temp 1(°c) | Wall temp 2(°c) | Outlet temp(°c) |
|-----------|----------------|-----------------|-----------------|-----------------|
| 0 | 50 | 37.6 | 38.9 | 41.9 |
| 2.5 | 44.9 | 34.5 | 36.6 | 38.2 |
| 5 | 40.8 | 32 | 33.9 | 35.2 |
| 7.5 | 37.4 | 30.3 | 32.1 | 32.7 |
| 10 | 34.8 | 28.6 | 30.2 | 30.5 |
| 12.5 | 32.5 | 26.1 | 26.8 | 28.7 |
| 15 | 30.5 | 24.9 | 25.2 | 27.2 |
| 15.51 | 30 | 24.5 | 25.1 | 26.8 |



Figure 15: Temperature vs Time for waterat 4lt/min

By comparing the readings and observations from table eleven to table thirteen, it is found that the heat transfer rate for water is max at a flow rate of 3.4lt/min.

6. Conclusion

The following conclusions have been made from the experiment MFrom the experiment, it has been found that the nano- coolant was most efficient at 0.15% volume concentration of TiO_2 in water at a flow rate of 3.4lt/min.

- 1. The least time taken by the nano fluid(TiO₂₊H₂O) is observed in table 6 i.e. 11 minutes. At the flow rate of 3.4l/min and 0.15% volume concenteration, the heat transfer coefficient, nusselt number is maximum.
- 2. The results for water as a coolant have been found maximum at a flow rate of 3.4lt/min.
- 3. The least time taken by water(conventional coolant) is observed in table 12 i.e. 13.5 minutes. It means water dissipates heat more effectively at a flow rate of 3.4 l/min.
- 4. By considering the above two points it is found that the results obtained by adding nano particle TiO_2 in water is more efficient than the conventional coolant water.

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